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Metal Ion-Promoted Syntheses of Boranes and Carboranes

Final Report

June, 1986

U. S. Army Research Office

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boron arene-metal complex' iron-carborane complex boranes cobalt-carborane complex carboranes nickel-carborane complex metallacarboranes oxidative fusion metal complex

20. ABSTRACT (Continue on reverse side if necessary and identity by block number)

Controlled synthetic routes to new boron hydrides, carboranes, and metal-carborane complexes have been developed and a number of structurally novel species have been prepared and characterized. The oxidative fusion method for synthesis of boranes and carboranes has been further extended and its mechanism explored. The chemistry of small (7-vertex) metallacarboranes having reactive metal centers has been examined. Practical routes to metalarene-carborane sandwich complexes have been developed.

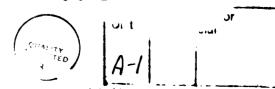
I. Statement of Pros em

This research was directed at the development of new synthetic routes to boron hydrides and carboranes. The work entailed fundamental studies in synthetic boron chemistry, centering on the use of transition-metal and main group-metal reagents to effect exidative fusion or coupling of small boranes and carboranes.

II. Summary of Important Results

1. Oxidative Fusion as a Synthetic Too!. The method of metal-promoted carborane case fusion, discovered in our laboratory in earlier work, was extended under the ARO Contract to boron hydrides and metallaboranes (Figure 1). This in turn led to a number of significant findings, including the synthesis and structural characterization of the first neutral B_{12} hydride ($B_{12}H_{16}$); the conversion of B_5H_9 to $B_{10}H_{14}$ via metal deprotonation and complexation; and the conversion of 1- and $2 - (C_5H_5)CoB_4H_8$ to isomers of (C_5H_5) $_2Co_2B_8H_{12}$, which are analogues of $B_{10}H_{14}$. The mechanism of fusion in these latter systems was investigated and the data were found to be consistent with base-to-base linkage of two-square-pyramidal units (B_5 or CoB_4) followed by rearrangement to a 10-vertex nido basket.

Further insight into the exidative fusion of $R_2C_2B_4H_4^{2-}$ ligands to form $R_4C_4B_8H_8$ (studied earlier for $R=CH_3$, C_2H_5 , and $In-C_3H_7$) was gained from the $R=C_6H_5CH_2$ case (C.C'-dibenzylcarborane, discussed below). The $(C_6H_5CH_2)_4C_4B_8H_8$



!es

product is non-fluxional, in contrast to its tetraalkyl analogues, and the case secmetry is the same as that in $(c_2H_5)_4c_4B_8H_8$.

- 2. Studies of (X)(Y)M($R_2C_2B_4H_4$) Metallacarboranes with Reactive Metal Centers [M = Fe, Co, Ni; X = Cl, Br, CN; Y = $P(C_6H_5)_3$, $(C_6H_5)_2PCH_2CH_2P(C_6H_5)_2$]. The preparation and chemistry of 7-vertex MC₂B₄ cases containing halo or cyano ligands was studied, as outlined in Figure 2. In addition to the chemistry showm, X-ray structural analyses of the diamagnetic species 1 and 2 and the paramegnetic complex 3 were conducted.
- 3. Synthesis and Chemistry of Arene-Metal-Carborane Sandwich Complexes. Methods have been developed for the synthesis of specific desired air-stable solid metallacarboranes in which the metal is π -coordinated to an arene ligand. Compounds of this type are of potential value in the areas of VHBR (very high burning rate) propellants, low-dimensional electrical conductors, and new reagents for organic synthesis. A wide variety of complexes containing the pyramidal $R_2C_2B_4H_4^{2-}$, cyclic planar $R_2C_2B_3H_5^{2-}$, or other carborane ligands, in conjunction with arenes and metals, has been prepared and structurally characterized. Figure 3 outlines the major developments in this chemistry under the ARO Contract.

Among the more important achievements in this work are the bench-scale synthesis of C₂C'-dibenzyldicarbahexaborane(8), an air-stable nonvolatile liquid which serves as a versatile precursor to numerous metal complexes (see Figure 3), and the designed preparation of a number of stable complexes containing metal-bound polyarene ligands including naphthalene, fluorene, phenathrene, and [2,2]paracyclophane. This work provides a foundation for continuing investigation in this laboratory.

FIGURE 1. OXIDATIVE FUSION IN SYNTHESIS

Pre-ARO work

*Maxwell, Miller, and Grimes, <u>J. Am. Chem. Soc.</u> <u>96</u>, 7116 (1974), and <u>Inorg. Chem.</u> <u>15</u>, 1343, 1976.

bWong, Bowser, Pipal, and Grimes, <u>J. Am. Chem. Soc</u>. 100, 5045 (1978)

OBH ●CR • H

CARBORANES

$$2\left[\begin{array}{c|c} & & & \\ \hline & & \\ \hline & & \\ \hline \end{array}\right] \xrightarrow{\text{FoCI}_2} \begin{array}{c} & & & \\ \hline & & \\ \hline & & \\ \hline \end{array} \begin{array}{c} & & \\ \hline & & \\ \hline \end{array} \begin{array}{c} & & \\ \hline & & \\ \hline \end{array} \begin{array}{c} & & \\ \hline & & \\ \hline \end{array} \begin{array}{c} & & \\ \hline & & \\ \hline \end{array}$$

R2C2B4H5

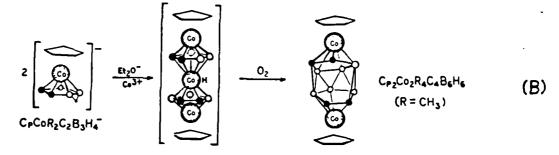
H2Fe(R2C2B4H4)2

R₄C₄B₈H₈

R=CH3, C2H5,

C₃H₇, CH₂C₆H₅

METALLACARBORANES



Work supported by ARO Contract

^aBrewer and Grimes, <u>J. Am. Chem. Soc.</u> <u>107</u>, 3552 (1985).

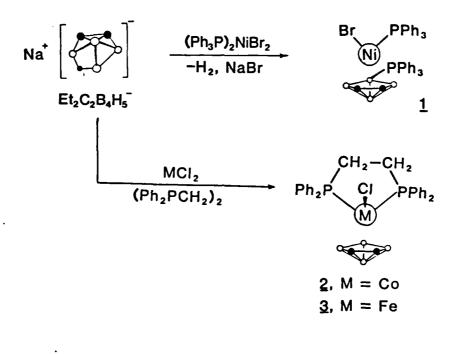
bBrewer, Swisher, Sinn, and Grimes, <u>J. Am. Chem. Soc</u>. <u>107</u>, 3558 (1985).

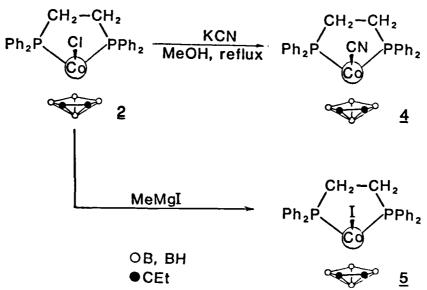
BORANES

METALLABORANES

CP2CO2BBHIZ ISOMERS

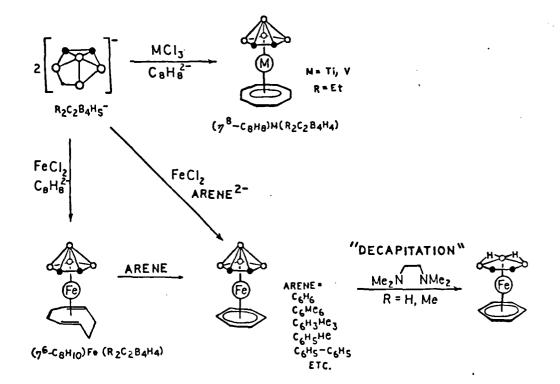
Figure 2. Synthesis and Chmistry of 7-Vertex $(X)(Y)M(R_2C_2B_4H_4)$ Metallacarboranes



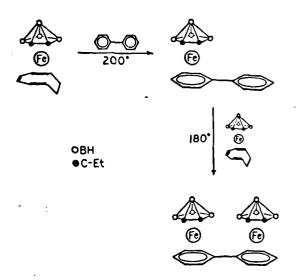


Synthesis of Arene-Metal-Carborane Sandwich Complexes

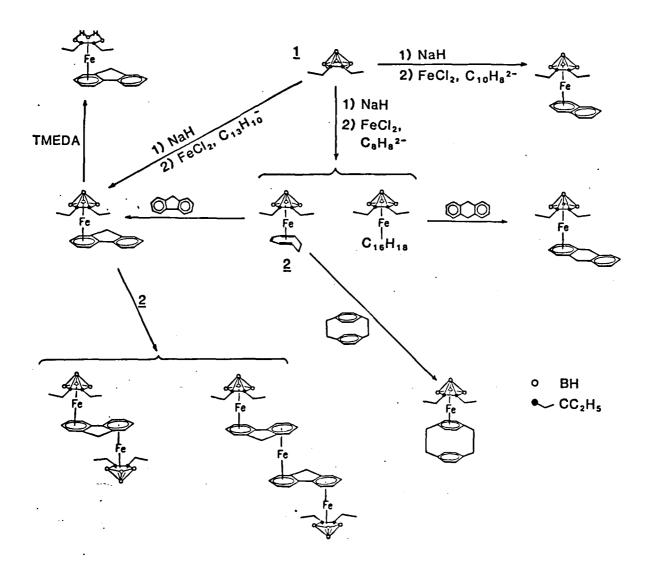
1. General Routes



2. Polycyclic Arene- and Multiidecker- Complexes



Polycyclic Arene- and Multiidecker- Complexes (cont'd)



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IV. Participating Scientific Personnel

Principal Investigator: Russell N. Grimes, Professor of Themistry

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Kathleen Kahler

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